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Enantiospecific Synthesis of 2-Amino-3-methyl-4 phosphonobutanoic Acids via 1,4=Addition of Lithiated Schöllkopf Anion to Prop-2-enylphosphonates

Vicente Ojea, María Ruiz, Gideon Shapiro* and Esteban Pombo-Villar^{*}

Preclinical Research, Sandoz Pharma Ltd, CH-4002 Basel.

Abstract: High diastereoselectivity in the conjugate addition of lithiated Schöllkopf's bislactim ethers to E- and Zprop-2-enylphosphonates was utilized to achieve a direct asymmetric synthesis of all four diastereoisomers of 2-
amino-3-methyl-4-phosphonobutanoic acid, i.e. $(+)$ - $(2S, 3R)$ -4a, $(+)$ - $(2S, 3S)$ -4b and their corresponding enantiomers. Their relative configuration was definitively assigned from an NMR study of oxaphosphinane derivatives of both diastereoisomers.

Conjugate additions of organometallic reagents to α , β -unsaturated carbonyl compounds constitute a powerful method which has found extensive use in stereospecific synthesis.¹ In contrast, analogous 1,4addition reactions to α , β -unsatured phosphoryl compounds have received little attention.² We have recently reported the synthesis of N-Fmoc-O,O-diallyl protected phosphonic acid analogues of phosphoserine (1a, *Fmoc-AbulPO(OAll)₂1-OH), via the coupling of the Schöllkopf anion (-)-2 with a bromoethylphosphonate. The* **unusually high diasterofacial selectivity in the coupling was attributed to a reaction mechanism involving** conjugate addition to a vinylphosphonate generated in situ.³

Figure 1 As part of a study directed toward the synthesis of the N-Fmoc-O,O-diallyl protected phosphonic acid analogues of phosphothreonine (1b, Fmoc-Ambu[PO(OAll)₂]-OH) we decided to investigate the conjugate addition of 2 to prop-2-enylphosphonates 3. This communication presents the first enantiospecific synthesis of all four diastereoisomers of 2-amino-3-methyl-4-phosphonobutanoic acid,⁴ (+)-(2S, $3R$)- $4a$, (+)-(2S, $3S$)- $4b$ and their enantiomers, by using a highly face-selective addition of the lithium salt of Schöllkopf's bislactim ether

(+)-2 or f-)-2 to E- and **Z-prop-2cnylphosphonatcs (34 and** *36,* **respectively).**

In spite of reports in the literature,⁵ Arbuzov reaction of excess 1,2-dibromopropane with triethylphosphite gave 5⁶ which was readily dehydrobrominated to 3a (Scheme 2). The Z-O,O-diethyl-prop-2-enylphosphonate 3b was prepared by Pd(0)-catalyzed coupling of cis-1-bromo-1-propene and diethylphosphite, as described in the literature.⁷

Schöllkopf described conjugate additions of lithiated bislactim ethers to α β -unsaturated carboxylic acid esters,⁸⁴ and of the corresponding cuprates to α , β -unsaturated ketones.⁸¹ These reactions were characterized by exceptionally high (d.e.>99%) diastereoselectivity at C-2 of the bislactim ether;⁹ however, variable diasteroselectivities at $C-1$, the β -carbon of the Michael acceptor, were observed. In the present study we found that the lithium salts of Schöllkopf's bislactim ether $(+)$ -2 or $(-)$ -2 rapidly add to both E and Z vinyl phosphonates $(3a$ and $3b$, respectively) with very high asymmetric induction for both chiral centers. As in the above Schöllkopf conjugate additions, the diasteroselectivity at C-2 was excellent $(d.e. > 95)$ in both cases, and in addition the diastereoselectivity at C-1' was also good (d.e.> 88%).¹⁰ Thus, the slow addition of 3a to (-)-2 at -78° C in THF followed by immediate quenching with acetic acid (-78 °C) and aqueous work-up gave a mixture of the Michael adducts $6a.6b$ in 94:6 ratio but in a combined yield of only 33% after chromatography. The low chemical yield is due to further capture of the initially formed Michael adduct anion by a second molecule of propenylphosphonate $3a$ to give $7a$, a mixture of diastereoisomeric 1:2 addition products. The isolated yield of the 1:1 adducts was improved to 86% simply by using excess (3 equivalents) $(-)$ -2 thereby suppressing formation of higher adducts. Addition of 3 equivalents of (-)-2 to Z-prop-2-enyl-phosphonate 3b proceeded with opposite stereoselectivity at C-1' to give 6a and 6b in 5:95 ratio and 84% isolated yield.

The separation of (+)-6a { $[\alpha]_D^2$ + 24.2 (CH₂Cl₂, c = 1)} from (-)-6b { $[\alpha]_D^2$ - 14.6 (CH₂Cl₂, c = 1.1)}could be achieved by medium pressure liquid chromatography (AcOEt-hexane, 1:1, SiO₂ 230-400 mesh) to provide products of high purity (d.e. > 98%) on a multigram scale. Mild acid hydrolysis of the bislactim ether provided the aminoesters $(+)$ -8a and $(+)$ -8b in excellent yield after chromatography (MeOH/AcOEt 10:1 after removing valine ester with AcOEt). Vigorous acid hydrolysis of these aminoesters followed by treatment of the crude hydrochlorides with propylene oxide and purification by reverse phase flash chromatography ¹¹ afforded the corresponding 2-amino-3-methyl-4-phosphonobutanoic acids $(+)$ - $(2S, 3R)$ -4a and $(+)$ - $(2S, 3S)$ -4b. Starting from $(+)$ -2, the corresponding enantiomers $(-)$ -4a and $(-)$ -4b were prepared in analogous fashion.¹²

Since none of the compounds 4-8 provided crystals suitable for an X-ray crystal structure determination. a cyclic derivative was sought which would enable assignment of the relative configurations of $4a$ and $4b$ by ¹H NMR spectroscopy. The six-membered amino-oxaphosphinane derivatives $I@a,b$ were attractive in this regard because of the the strong preference for chair conformation reported for similarly 4-methyl-substituted systems.¹³ Chemoselective reduction of the carboxylic ester in the presence of the phosphonic ester was eventually achieved on the N-Boc derivatives of $(-)$ -8a and $(-)$ -8b with LiBH₄ in anhydrous THF at RT.¹⁴ Under these conditions, cyclization of the reduction intermediates to afford the corresponding N -Boc-aminooxaphosphinane derivatives $(+)$ -9a (70%) and 9b (69%) was observed.

 i : 1.2 eq (BOC)₂O, 0.8 eq Na₂ CO₃, 0.9 eq NaHCO₃, dioxane:H₂O 1:1, RT, 3 h, 90-95%;
 ii : 1.5 eq LIBH₄, THF, RT, 16 h, 70-75%; iii : 2.3 eq TMSBr, CH₂CI₂, RT, 14 h, 80-84% **Scheme 4**

Compound (+)-9a was obtained as one single diastereoisomer {mp (AcOEt) = 121-124 °C, $\left[\alpha_{\text{in}}\right]^2$ + 74.8 $(CH₂Cl₂, c = 1)$, while 9b was isolated as a mixture of isomers at the phosphorus center in 2.7:1 ratio which could not be separated by simple chromatography.¹⁵ The Boc and ethyl ester groups of $(+)$ -9*a* and 9*b* were cleaved with TMSBr in CH₂Cl₂ to give (+)-10a { $[\alpha]_D^{\infty}$ = + 35.6 (H₂O, c = 1)} and (+)-10b {mp (H₂O) = > 200 °C (dec.); $[\alpha]_n{}^{2} = +7.6$ (H₂O, c = 0.95)} as pure diastereoisomers. A single chair conformation was observed for each compound 10 in the ¹H NMR spectra (D₂O, 400 MHz, 25 °C). For compound (+)-10b a triple doublet was observed for H-5 at δ = 3.24 with coupling constants J = 9.1, 3.1 Hz. The coupling constant of 9.1 Hz indicated a diaxial relationship between H-5 and H-4 with degenerate coupling of H-5 to H-6ax. Futhermore, the complete set of NOE cross peaks that were observed by ¹H-¹H ROESY ¹⁶ was consistent with this trans-stereochemical assignment (see Fig. 2). The assignment of the cis-configuration for diastereoisomer $(+)$ -10a follows by default and is also consistent with the NOEs observed (see Fig. 2). The absolute configuration was then assigned under the assumption that the configuration at C-5 is determined by the configuration of the valine residue in the bislactim ether, as there is ample precedent.

Figure 2. Selected observed NOE's.

The stereochemical course of these conjugate additions to vinyl phosphonates is similar to that previously encountered in the case of α B-unsaturated carboxylic esters;³⁴ thus the selective formation of the Michael adducts $(2S, 1R)$ -6a in the reaction of (-)-2 with the E-prop-2-enylphosphonate and $(2S, 1S)$ -6b when (-)-2 reacted with Z-prop-2-enylphosphonate can be understood as a consequence of almost exclusive interaction of the Si-face of $(-)$ -2 with the Si-face of 3a and the Re-face of 3b. Presumably, lithium chelation and interactions between n-systems, as was suggested by Schöllkopf,^{3a} are operating similarly in this case. However, the higher stereoselectivity observed in the present work may be accounted for by the different steric and electronic interactions of the tetrahedral phosphoryl group.

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Chromatography through a short column (Merck LiChroprep RP-18, 40-63 µ) using H₂O as eluent afforded amino acids
- 11. 4a,b almost with the solvent front, other impurities being retained.
- 12. (+)-(2S, 3R)-4a; mp (EtOH) = >80 °C (dec.); [a]_D²⁰ = + 31.8 (H₂O, c = 1); ¹H NMR (D₂O, 360 MHz): 1.17 (1H, d, J = 7.2 Hz, CH₃), 1.77 (1H, ddd, J = 9.1, 15.0, 18.1 Hz, H-4), 1.82 (1H, ddd, J = 6.1, 15.0, 19.2 Hz, H-4), 2.52 (1H, m, H-3), 4.09 (1H, d, J = 3.6 Hz, H-2); ¹³C NMR (D₂O, 90 MHz); 18.7 (d, J = 6.4 Hz, CH3), 33.1 (C-3), 33.8 (d, J = 128 Hz, C-4), 61.1 (d, J = 12.7 Hz, C-2), 174.3 (C-1). (+)-(2S, 3S)-4b: mp (EtOH) = >80 °C (dec.); [a]_D^D = + 14.2 (H₂O, c = 1); ¹H NMR (D₂O, 360 MHz): 1.15 (1H, d, J = 7.1 Hz, CH₃), 1.73 (1H, ddd, J = 23.1, 15.3, 7.8 Hz, H-4), 1.87 (1H, ddd, J = 18.7, 15.3, 6.1 Hz, H-4), 2.60 (1H, m, H-3), 4.08 (1H, d, J = 3.4 Hz, H-2); ¹³C NMR (D₂O, 90 MHz); 18.6 (d, J = 7.6 Hz, CH3), 32.9 (C-3), 33.7 (d, $J = 132$ Hz, C-4), 61.2 (d, C-2), 174.7 (C-1).
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